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## Modeling of laser-induced modifications of nano-objects and nanocomposite materials

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## **Nanoparticles and Nanocomposites**



### **Car industry**



#### **Aerospace and Defense**



#### Z. Liu, et al. JPC C 119.17 (2015): 9496-9505.

#### Sensors and **Catalysis**



## Images and multiplexing (collab. with N. Destouches et. al.)

#### fs, ns or CW laser



J. Mater. Chem. C 2.31 (2014;

Nanotechnology 28.3 (2016): 035302.

#### multiplexing



*Z. Liu et al., ACS nano 11.5 (2017): 5031-5040. N. Sharma et al., Materials Horizons 6.5 (2019): 978-983.* 

## **Laser-Assisted Nanotechnology**



In gas or in vacuum with different geometry



(c)

In liquids or in the precence of a colloids



## Among main advantages

- Green synthesis
- Non-toxic, bio-compatible
- Environment friendly

### Laser allow

- -Control over NP properties
- -Scaling-up possibilities
- -Possibility to create stable ultrastructures, nanoalloys, hybrids, etc. -Additional laser-particle and lasernanocomposite interactions to produce new properties (sizes, colors, optomagnetic properties, etc.).

## **Ultra-short Laser Interactions with NPs**



Lachaine et al. Photonics (2013)



#### Hashimoto et al. J. PhotChem&PhotBiology (2012)



Strasser et al, . J. Phys.Chem. C 118, 25748 (2014)

#### **NP fragmentation**

## Early MD Simulations (collab. with L. Zhigilei)

# Decomposition of the ablated material after pulsed laser irradiation and additional NP's fragmentation



TE Itina, LV Zhigilei, BJ Garrison The Journal of Physical Chemistry B, 106(2), 303, 2002 t = 300 ps τ = 15 ps e=0.6 eV per molecule

T.E. Itina, L. V. Zhigilei, B.J. Garrison, Nuclear Instruments and Methods in Physics Research Section B, 180(1-4), 238-244, 2001

## **MD** simulations: Role of NP size



#### Smaller NPs

#### **Evaporation**

but can be also due to Coulomb Explosion for small clusters and fs laser, etc..

#### R=150, absorbed energy e=0.6 eV/mol.



#### Larger NPs

# Fragmentation due to thermo-mechanical effects

and/or instabilities

## **NP's Heating and Melting : Size Effects**

Melting and structural transformations: role of NP's sizes



K. Zarschler et al. Nanomed-Nanotechnol 12, 1663-1701 (2016).





## Laser Energy Absorption vs NP's size

## **Absorption coefficient / efficiency**



## Laser Interaction with NP in Water (collab. with A. Rudenko)

#### **Classical 3T models** Au NP in water



 $C_e, C_{NP}, C_m$  - heat capacities;  $k_e, k_{NP}, k_m$  - thermal conductivities;  $T_e, T_{NP}, T_m$  - temperatures;  $\vec{J} \cdot \vec{E}$  - Joule heating source (absorption);  $\gamma_{ei}$ , h - energy transfer rates;  $\rho_{NP}$ ,  $\rho_m$  - ion densities.



### Advantages: rather easy equations, estimations of temperatures Limitations: hard to correctly account for electron excitation, ejection/injection

## **Ultra-Short Laser Interaction with Metallic NPs**

## **Non-Equilibrium Electron Sub-system**

#### **Conduction band (free) electrons in metals**

Boltzmann equation

$$\frac{\partial f(E)}{\partial t} = \left(\frac{\partial f(E)}{\partial t}\right)_{e-e} + \left(\frac{\partial f(E)}{\partial t}\right)_{e-ph} + \left(\frac{\partial f(E)}{\partial t}\right)_{intra/inter}^{abs}$$

$$\left(\frac{\partial f(E)}{\partial t}\right)_{e-e} \cong \frac{f(0)C_{ee}}{\sqrt{E}} \frac{\partial}{\partial E} \left[f(E)\left(1-f(E)\right) + k_B T_e \frac{\partial f}{\partial E}\right]$$

$$\left(\frac{\partial f(E)}{\partial t}\right)_{e-ph} \cong \frac{C_{ep}}{\sqrt{E}} \frac{\partial}{\partial E} \left[ f(E) \left(1 - f(E)\right) + k_B T_i \frac{\partial f}{\partial E} \right]$$

 $\left(\frac{\partial f(E)}{\partial t}\right)_{\substack{abs\\intra/inter}} \cong \frac{P_{abs(t)}}{\kappa} [f(E-\hbar\omega) - 2f(E) + f(E+\hbar\omega)]$ 

Definitions of electron & lattice temperatures

 $E_{max} = E_F + \varphi$ 

$$k_B T_e = \frac{1}{f(0)} \int_0^{E_{max}} f(E) (1 - f(E)) dE$$

$$\rho_{NP}C_{NP}\frac{\partial T_{NP}}{\partial t} = -A \int_{0}^{E_{max}} \left(\frac{\partial f(E)}{\partial t}\right)_{e-ph} E^{3/2} dE$$

Here, we can transit back to spatial TTM model to add spatial thermal diffusion etc.

## **Ultra-Short Laser Interaction with Metallic NPs**

#### Can we account for electron emission/injection into ambient medium ?

## At metal-water interface

#### Nonequilibrium approach

Thermionic photo-emission (electrons with energies exceeding  $E_{max}$  leave metal)

$$\begin{pmatrix} \frac{\partial N_e}{\partial t} \end{pmatrix}_e = AC_{ee}k_B T_e \left( \frac{\partial f(E)}{\partial E} \right)_{E=E_{max}}$$

$$\begin{pmatrix} \frac{\partial N_e}{\partial t} \end{pmatrix}_p = AC_{ep}k_B T_i \left( \frac{\partial f(E)}{\partial E} \right)_{E=E_{max}}$$

$$\begin{array}{c} \frac{\partial N_e}{\partial t} = -\frac{1}{e} \nabla \vec{J} \end{array}$$

#### Photoelectric effect/Photo-emission

$$\left(\frac{\partial N_e}{\partial t}\right)_{ph} = A \int_{E_{max}-h\nu}^{E_{max}} \left(\frac{\partial f(E)}{\partial t}\right)_{intra} \sqrt{E} dE$$

#### Fowler formulas for current density

Generalized Fowler-DuBridge theory of photoemission

$$J_n(T_e) = a_n \left(\frac{e}{h\nu}\right)^n I^n T_e^2 F\left(\frac{nh\nu - \varphi}{k_B T_e}\right)$$

Thermionic emission (Richardson law)

$$J(T_e) = A_0 T_e^2 exp\left(-\frac{\varphi}{k_B T_e}\right)$$

Fowler-Nordheim theory for field emission (threshold occurs to be above the optical breakdown

in water for ultrashort pulse)

$$J(E) = \frac{a}{\varphi} E^2 exp\left(-\frac{v(E)b\varphi^{3/2}}{E}\right)$$

## Au NPs in Water (collab. with A. Rudenko)

fs laser  $\lambda = 515 \text{ nm}$ 

## What changes when nanoparticle size inceases ?



## Several Interaction Regimes (collab. with A. Rudenko)

#### What changes when nanoparticle size inceases ?



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## **Fragmentation: MD simulations**



In situ observation of gold nanoparticle fragmentation under femtosecond laser irradiation



#### Gold nanoparticles in water

*Experiments: G. Bongiovanni et al.* Nanoscale Adv., 2021, 3, 5277-5283 Our MD results for NPs of different materials, absorbed energy, and sizes



## Metal alloy NPs: Role of the absorbed energy per atom

Au-Co NP , R= 40 u.cells (d~30 nm) 50%, t= 0, 150, 400 ps absorbed energy below the threshold



Au-Co NP, 50% R=40 uc t= 0, 150, 400 ps, absorbed energy above the threshold



## Metal Alloy NP : fragmentation, MD results

**Au-Co NPs,** 50% R= 40 u.c. t= 0, 200, 1600 ps The absorbed energy/atom is right above NP fragmentation threshold



Rather homogenous decomposition and numerous small fragments are observed

## Metal Alloy NP : fragmentation, MD results

**Au-Ni NP,** 50% R=40 t= 0, 200, 1600 ps, The absorbed energy/atom is right above NP fragmentation threshold

Here, the absorbed energy/atom is 25% smaller for AuNi then for AuCo



**Decomposition to only several fragments** 

## Many Laser Pulses: Additional Fragmentation

AuCo, NP R = 40 u.c. Simulat

Simulations starts from the fragmentation results, additional fragmentation takes place



Upon add fragmentation, re-organization and cooling



Reducing the size of absorbing fragments

## **Ion-impregnated Porous Thin Films Treated by Laser**

- sol-gel mesoporous films are prepared and impregnated by saults

- these samples are then irradiated by a laser

Porous films:  $SiO_2$ ,  $TiO_2$ 

Sol-gel synthesized Porosity ~ 10% Porous size 5-7 nm

## Metallic NPs:

Ag, Au

Produced by soaking in metallic salts solutions with consequent reduction Initial size < 3-5 nm Laser irradiation: CW Nanosecond Picosecond Femtosecond







Ion-impregnated porous glasses and semiconductors can be used as samples for following laser-induced nano-structuring

## Nanoparticle formation and growth



#### 2. Heating and thermal diffusion



## 5. Photo-oxydation



### 3. Photoemission and reduction



H. Ma et al. *J. Phys. Chem. C* 123.42 (2019): 25898-25907

#### 6. NPs growth



Multi-scale and multi-physical mechanisms are involved.

## Au NPs in TiO<sub>2</sub> under Ultra-Short Laser Irradiation





(b)Calculated absorption efficiency of the nanocomposite based on TiO2 as a function of the diameter of gold NPs

(c) Calculated fileId distribution of gold nanoparticles in TiO2 (n=1.6) for spheres with the diameters of 50 and 90 nm.

(d) SEM images of the NPs produced in LS mode (F = 57 mJ/cm2, Vsc = 0:16 mm/s).

## Au NPs in TiO<sub>2</sub> under Ultra-Short Laser Irradiation



#### Electron (black solid) and ion temperature (red solid) of R=45 nm gold NP and titanium dioxide temperature (blue solid)

Here, additionally includes electron photo-emission.

The Journal of Physical Chemistry C 124 (18), 10209-10219, 2020

It is important to account for collective heating of NP ensemble with bimodal size distribution, for electron photoemission from nanoparticles and free carrier generation in titanium dioxide to reach the damage conditions



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## **Summary and Conclusions**



- Laser is a very powerful and versatile tool, which is particularly suitable for many applications in the modern nanotechnology.
- Particle size, composition, geometry, environment, absorption and scattering, heating/cooling protocols strongly affect the results of laser interactions.

-NPs of smaller sizes easily melt and evaporate, while larger NPs behave more like bulk material, so that thermo-mechanical fragmentation of over-heated material plays a role in their fragmentation. Phase diagrams of NPs smaller than ~30-100 nm should account for NP's size.

-Fragmentation threshold of alloy nanoparticle strongly depends on the material properties and composition.

• Optical properties are strongly related to size and distribution of NPs after laser treatment, are controlled by laser processing parameters. 24







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# Thank you very much!

Just published:

"Alternative local melting-solidification of suspended nanoparticles for heterostructure formation enabled by pulsed laser irradiation" Advanced Functional Materials 2023.

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